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Brønsted Base-Modulated Regioselective Pd-Catalyzed Intramolecular Aerobic Oxidative Amination of Alkenes: Formation of Seven-Membered Amides and Evidence for Allylic C—H Activation

Liang Wu, Shuifa Qiu, and Guosheng Liu*

State Key Laboratory of Organometallic Chemistry, Shanghai Institute of Organic Chemistry, Chinese Academy of Sciences, 354 Fenglin Road, Shanghai, China, 200032 gliu@mail.sioc.ac.cn

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ABSTRACT

$$\begin{array}{c} \text{BzONa} \\ \text{(1-2 equiv)} \\ \text{additive} \\ \text{R = H, Me} \\ \text{NHTs} \\ \text{DMA, O}_2 \\ \\ \text{NHTs} \\ \text{DMA, O}_2 \\ \\ \text{R = H} \\ \text{without base} \\ \text{additive} \\ \\ \text{R = H} \\ \text{NHS} \\ \text{$$

A novel palladium-catalyzed intramolecular aerobic oxidative allylic C—H amination of olefins has been developed. Brønsted base can modulate the regioselectivity, favoring the formation of 7-membered rings. Mechanistic studies using deuterium-labeled substrates as probes support a rate-determining allylic C—H activation/irreversible reductive elimination pathway.

The prominence of nitrogen-containing heterocycles in natural products and biologically active molecules has prompted considerable efforts toward their synthesis. Among transition-metal promoted cyclizations of alkenes, palladium-catalyzed oxidative aminations represent one of the most efficient synthetic routes for nitrogen-containing heterocycles. An even more desirable approach for catalytic C-N bond formation is the dioxygen-coupled oxidative amination of olefins. While formations of five- and six-membered heterocycles have dominated palladium-catalyzed oxidative intramolecular aminations of alkenes, formations of seven-

membered nitrogen-containing heterocycles under oxidative conditions are quite rare. Herein, we report a Brønsted base-modulated regioselective Pd-catalyzed intramolecular aerobic oxidative amination of alkenes for seven-membered amides

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formation.^{5,6} Mechanistic studies support the allylic C-H activation pathway.

Recently, significant progress has been made in the area of metal-catalyzed amination of C–H bonds; 7,8 for instance, Rh-catalyzed C–H amination through highly reactive metallo-niteneoids, $^{7a-d}$ Pd- or Cu-mediated C–H oxidative amination assisted by directing groups, $^{7e-g,8}$ and Pd-catalyzed allylic C–H oxidative amination. 6b,c,9 The latter have proposed a π -allylpalladium species as key intermediates. For intramolecular cyclizations, since the π -allylpalladium intermediate normally exists as an equilibrium mixture of syn (I-A) and anti (I–B and I–C), the subsequent nucleophilic attack may afford five- and/or seven-membered rings depending on whether the attack proceeds through exo or endo fashion (Scheme 1). We hypothesized that Brønsted

Scheme 1. Pd^{||}-Mediated Intramolecular Oxidative Allylic C-H Amination of Alkenes

base can promote the Pd-N bond formation¹⁰ to favor intermediate **I**-**C**, leading to selective generation of seven-membered rings (Scheme 1, bottom).

Our experiments were initiated with the intramolecular allylic amination reaction of monosubstituted olefine **1a**. The reaction mediated by Pd (OAc)₂ under aerobic condition yielded 20% five-membered ring product **3a** with high regioselectivity (Table 1, entry 1). To our surprise, the seven-

Table 1. Pd-Catalyzed Intramolecular Oxidative Amination of **1a**^a

$$\begin{array}{c|c} O & Pd(OAc)_2 \ (10 \ mol \ \%) \\ \hline NHTs & additive \\ DMA, O_2 \ (1 \ atm) & Ts \\ \hline \mathbf{2a} & \mathbf{3a} \\ \end{array}$$

entry	additive	yield (2a:3a) ^b
1	none	20% (<3:97)
2	NaOBz (100 mol %)	44% (82:18)
3^c	NaOBz (100 mol %), 4 Å MS (15 mg), MA	
	(40 mol %) [Condition A]	91% (86:14)
4	(salen)Cr(III)Cl (10 mol %) [Condition	
	B]	65% (<3:97)
5	NaOBz (100 mol %), (salen)Cr(III)Cl (10	
	mol %)	54% (80:20)

 $[^]a$ Reactions were conducted at 0.1 mmol scale at 1 mL DMA under 1 atm dioxygen, 50 °C. b ¹H NMR yield, 1,3,5-trimethoxybenzene as internal standard. c At 70 °C.

membered ring **2a** was formed as the major product in the presence of a base additive (NaOBz, 1 equivalent) under otherwise identical conditions (entry 2). The formation of **2a** can be further enhanced by adding maleic anhydride (MA, 40 mol %) and 4 Å molecular sieves (condition A, entry 3). On the other hand, **3a** can also be obtained in good yield by adding (salen)Cr^{III}Cl, an additive first reported by White and co-workers (condition B, entry 4). However, when base is added to the system containing (salen)Cr^{III}Cl, the regioselectivity of the reaction was reverted with the generation of 7-membered ring **2a** again (entry 5). The above results indicated that the use of base additive is crucial for the formation of 7-membered ring, as proposed in Scheme 1. Is

The base-promoted regioselective formation of 7-membererd rings is general with respect to monosubstituted alkenes (Table 2). For substrates 1a-1f tested in our experiments, the optimal condition described above (condition A) favored 7-membered rings 2a-2f formation (selectivity ranging from 79:21 to 86:14, odd entries). In cases where no base was added (Condition B), 5-membered rings 3a-3f were formed selectively (even entries). It is worth noting that reactions of 1b-1e exhibited high stereoselectivity to afford trans products 3b-3e (entries 4, 6, 8 and 10).

The regioselectivity of 7- over 5-membered rings may be further enhanced by using substrates with a methyl group installed at an internal vinyl position. This is because the

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Table 2. Palladium-Catalyzed Oxidative Allylic C-H Amination^a

entry	substrate	condition	major	product	yield $(2:3)^b$
1	1a $R^1 = H, R^2 = H$	A	2a		88% (86:14)
2	1a	В		3a	61% (<3:97)
3	1b $R^1 = Me, R^2 = H$	A	2b		83% (82:18)
4	1b	В		3b $(10:1)^c$	66% (<3:97)
5	$1c R^1 = C_5 H_{11}, R^2 = H$	A	2c		64% (81:19)
6	1c	В		$3c (6:1)^c$	71% (6:94)
7	1d $R^1 = {}^{i}Pr, R^2 = H$	A	2d		56% (81:19)
8	1d	В		3d $(9:1)^c$	41% (<3:97)
9	$1e R^1 = CH_2OBn, R^2 = H$	\mathbf{A}	2e		67% (79:21)
10	1e	В		$3e (17:1)^c$	61% (5:95)
11	$\mathbf{1f} \; \mathrm{R}^1 = \mathrm{Me}, \; \mathrm{R}^2 = \mathrm{Me}$	A	2f		74% (84:16)
12	1f	В		3f	71% (<3:97)

^a Reaction was conducted at 0.2 mmol scale at 2 mL DMA under 1 atm dioxygen. Condition **A**: Pd(OAc)₂ (10 mol %), NaOBz (1 equiv), MA (40 mol %), 4 A MS (30 mg), 70 °C; Condition **B**: Pd(OAc)₂ (10 mol %), (salen)Cr(III)Cl (10 mol %), 50 °C. ^b Isolated yield, the data in parentheses is the ratio of **2** and **3**. ^c Ratio of *trans*-3 and *cis*-3.

(entries 14-15).

Amination of 4^a

entry

substrate

reaction of 4a is expected to favor π -allylpalladium intermediates \mathbf{H} over $\mathbf{H}\mathbf{H}$ due to the higher activity of the terminal methyl over the internal methylene group. ¹⁴ The intermediate \mathbf{H} should exclusively give seven-membered product $\mathbf{5a}$ (eq 1). We found when substrate $\mathbf{4a}$ was submitted to condition \mathbf{A} , similar ring-sized selectivity ($\mathbf{5a/6a} = 87:13$) was achieved as in the cyclization of $\mathbf{1a}$ (eq 1). The fact that $\mathbf{5a'}$ was not observed indicated that intermediate $\mathbf{H}\mathbf{H}$ afforded five-membered ring $\mathbf{6a}$ as the only product. ¹⁵ However, for the substrate $\mathbf{4a}$, no reaction occurred under condition \mathbf{B} .

Subsequent studies of substrate scope were conducted. Interestingly, for substrates $\bf 4b-4i$, which featured one or two substituents at β -position of carbonyl groups, the reactions afforded the seven-membered products $\bf 5b-5i$ in good yields and excellent regioselectivities (Table 3, entries 2–10). This can be attributed to complete elimination of the activation of allylic C–H bond of the methylene group probably due to steric effect. Product $\bf 5f$ was successfully prepared on a 1.5 mmol scale in yield comparable to that on small scale (entry 7). The seven-membered rings $\bf 5j-5l$ containing two hetereoatoms were also prepared using our methods in moderate yields (entries $\bf 11-13$). Finally, for the one methylene less substrates $\bf 4m$ and $\bf 4n$, the reactions

yields and excellent regioselectivities (Table 3, entries

proceeded smoothly to six-membered rings with good yields

reaction may proceeds through an allylic C-H activation/

reductive elimination pathway, or an aminopalladation/ β -H

elimination pathway. The deuterium labeled experiments can

be used to differentiate the above two scenarios. If the

reaction involves allylic C-H activation, both the reactions

yield^b

product

Table 3. Palladium-Catalyzed Oxidative Allylic C-H

We then conducted preliminary mechanistic studies. The

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^{71%} 10 4a (87:13)^f $R^1 R^2$ NHTs 2^d **4b** R^1 = H, R^2 = Me 76% 3¢ 4c R1 = H, R2 = Ph 58% **4d** R¹ = H, R² = OMe 5d 82% 4 5 6 4e R1 = H, R2 = COOMe 5e 70% 5f 4f R^1 = Me, R^2 = Me 91% 4f 5g $4a R^1 = Me. R^2 = OMe$ 9 **4h** $R^1 = R^2 = (CH_2)_4$ 82% 10 4i $R^1 = R^2 = (CH_2)_5$ NHTs 4j X = 0, R = H 11 54% 4k X = NTs, R = H 12 62% 51 13 41 X = NTs, R = Me 68% 4m R = H 4n R = Me 14 15 72% 77%

 $[^]a$ Reaction condition: **4** (0.2 mmol), Pd(OAc)₂ (10 mol %), NaOBz (1 equiv), MA (40 mol %), 4 A MS (30 mg) in DMA (2 mL) under 1 atm O₂, 70 °C. b Isolated yield. c NaOBz (2.5 equiv). d NaOBz (1.5 equiv), 20 h. e Scale: 1.5 mmol. f Ratio of **5a:6a**.

⁽¹¹⁾ For the screening results, see Supporting Information Table S1.

⁽¹²⁾ Covell, D. J.; White, M. C. *Angew. Chem., Int. Ed.* **2008**, *47*, 6448 (Salen = 1,2-cyclohexanediamino-*N*,*N'*-bis(3,5-di-*tert*-butyl-sali-cylidene).

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⁽¹⁵⁾ The isomerization from **5a'** to **5a** is unlikely due to relative thermodynamic stability, see: Eskola, P.; Hirsch, J. A. J. Org. Chem. **1997**, 62, 5732. On the other hand, no observation of the double bond isomers within the recovered **4a**-A in eq 2also excludes this possibility.

⁽¹⁶⁾ The product **5f** was treated with Na/Naphthylene and hydrogenation to afford 4,4,6-trimethyl carprolactam in 85% yield.

of **4a**-A and **4a**-B will result the same π -allyl-Pd^{II} intermediate **II**-d and afford the similar results containing the mixture of equal amounts of **5a**-A and **5a**-B (with negligible secondary isotope effect, see Scheme 2, path a). Otherwise,

Scheme 2. Hypothesis of Two Possible Mechanisms (AMP = Aminopalladation)

if the reaction goes through aminopalladation/ β -H elimination pathway, the reactions of **4a-A** and **4a-B** would be formed seven-membered products **5a-A** and **5a-B** respectively (path b).

The reactions with D-labeled substrate 4a-A afforded a mixture of **5a-A** and **5a-B** with a ratio of 48:52 in low yield (16%), and the five-membered product $\mathbf{6a-}A$ (10%)¹⁷ (eq 2). Substrate 4a-A was recovered in 52% yield without alkene isomerization. For the substrate 4a-B, the reaction also produced the mixture of 5a-A and 5a-B with the ratio 46:54 in 61% yield, and **6a-B** in 9% yield (eq 3). The formations of mixtures of 5a-A and 5a-B with roughly 1:1 ratio in both reactions exclude the aminopalladation/ β -H elimination pathway. Furthermore, the lower conversion of 4a-A is probably resulted from a higher energy barrier of the allylic C-D activation. The kinetic isotopic effect (KIE) of allylic C-H activation was obtained with a 1:1 mixture of 4a and **4a-A**, and the large primary KIE value $(k_H/k_D = 5.75)$ supports our hypothesis of a turnover-limiting allylic C-H activation pathway (eq 4).18,19

In conclusion, we have developed a new palladium-catalyzed intramolecular aerobic oxidative allylic amination of unactivated olefins. Brønsted base can modulate the regioselectivity, favoring the formation of 7-membered products. This methodology provides an efficient route for the synthesis of seven-membered cyclic amides. Mechanistic studies support that the reaction proceeds through a rate-determining allylic C-H activation/irreversible reductive elimination pathway to form the C-N bond.

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Supporting Information Available: Experimental procedures and characterization data for all new compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

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⁽¹⁷⁾ The formation of **6a-A** also supports the allylic C-H activation pathway, see Supporting Information Scheme S2 for details.

⁽¹⁸⁾ For the stoichiometric reaction of PdCl₂ and methylenecyclo-hexane, the *KIE* values (4.55–5.45) of allylic C–H activation were observed, see: (a) Chrisope, D. R.; Beak, P. *J. Am. Chem. Soc.* **1986**, *108*, 334. (b) Chrisope, D. R.; Beak, P.; Saunders, W. H. *J. Am. Chem. Soc.* **1988**, *110*, 230.

⁽¹⁹⁾ The formation of seven-membered product 2a is the kinetically preferred product formation and the C-N bond formation is irreversible step, for more detail see Supporting Information.